organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Xiao-Hua Zeng, Hong-Mei Wang, Zai-Gang Luo, Ming-Wu Ding and Hong-Wu He*

Key Laboratory of Pesticide and Chemical Biology of Ministry of Education, College of Chemistry, Central China Normal University, Wuhan 430079, People's Republic of China

Correspondence e-mail: zengken@126.com

Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(C-C) = 0.005 \text{ Å}$ Disorder in main residue R factor = 0.060 wR factor = 0.155 Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-(2-Aminophenyl)-2-methylsulfanyl-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidin-4(3*H*)-one

In the title compound, $C_{17}H_{17}N_3OS_2$, the fused thiophene and pyrimidine rings are almost coplanar. The amino group is involved in both intramolecular and intermolecular hydrogen bonds, the latter linking the molecules into a centrosymmetric dimer.

Comment

Pyrimidine derivatives are increasingly attracting the attention of the synthetic community because of the important role played by such systems in many natural products, antibiotics and drugs (Ding *et al.*, 2004). In this paper, we present the X-ray crystallographic analysis of the title compound, (I).



The molecular structure indicates that the thieno[2,3-d]pyrimidine group is a conjugated system (Fig. 1). The tetrahydrobenzene ring is disordered. There are two possible conformations, C1–C6 and C1/C2'/C3'/C4–C6, with an occupancy ratio of 0.554 (8):0.446 (8). The amino group of the



View of the molecule of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at 50% probability level. Both disorder components are shown.

O 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

Received 31 October 2005 Accepted 10 November 2005 Online 19 November 2005 molecule forms weak hydrogen bonds with atoms N2 and O1 (Table 2 and Fig. 2). The former is an intramolecular hydrogen bond, which completes a closed five-membered loop. The latter is an intermolecular interaction with a neighbouring molecule related by a centre of inversion. The intermolecular interactions thus link the molecules into a hydrogen-bonded dimer.

Experimental

A mixture of 3-(2-aminophenyl)-2-thioxo-2,3,5,6,7,8-tetrahydrobenzothieno[2,3-d]pyrimidin-4(1H)-one (1 mmol), methyl iodide (1.3 mmol) and solid potassium carbonate (2 mmol) in CH₃CN (10 ml) was stirred for 1 h at room temperature and filtered. The filtrate was concentrated and the precipitate was recrystallized from methylene chloride/petroleum ether (1:3 v/v) to give compound (I) in 80% yield (m.p. 507 K). ¹H NMR (CDCl₃, 400 MHz): δ 7.33–6.89 (m, 4H, Ar-H), 3.34 (br, 2H, NH₂), 2.96-2.75 (m, 4H, 2CH₂), 2.48 (s, 3H, SCH₃), 1.91–1.80 (*m*, 4H, 2CH₂); MS (*m*/*z*, %), 343 (M^+ , 100), 326 (18), 295 (65), 267 (27), 254 (7), 117 (6). Analysis calculated for C17H17N3OS2: C 59.45, H 4.99, N 12.23%; found: C 59.68, H 4.92, N 12.35%. Crystals suitable for single-crystal X-ray diffraction were obtained by recrystallization from a mixed solvent of hexane and dichloromethane (1:3 v/v) at room temperature.

Crystal data

$C_{17}H_{17}N_3OS_2$	$D_x = 1.388 \text{ Mg m}^{-3}$
$M_r = 343.46$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 1263
a = 11.7686 (10) Å	reflections
b = 13.4861 (11) Å	$\theta = 2.4 - 19.6^{\circ}$
c = 11.2137 (10) Å	$\mu = 0.33 \text{ mm}^{-1}$
$\beta = 112.552 \ (2)^{\circ}$	T = 292 (2) K
V = 1643.7 (2) Å ³	Block, colourless
<i>Z</i> = 4	$0.30 \times 0.20 \times 0.10 \ \text{mm}$
Data collection	

Bruker SMART CCD area-detector	2125 reflections with $I > 2\sigma(I)$
diffractometer	$R_{\rm int} = 0.054$
φ and ω scans	$\theta_{\rm max} = 27.0^{\circ}$
Absorption correction: none	$h = -14 \rightarrow 15$
9565 measured reflections	$k = -14 \rightarrow 17$
3575 independent reflections	$l = -12 \rightarrow 14$

Refinement

Refinement on F^2	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.061$	independent and constrained
$wR(F^2) = 0.155$	refinement
S = 0.99	$w = 1/[\sigma^2(F_o^2) + (0.0702P)^2]$
3575 reflections	where $P = (F_0^2 + 2F_c^2)/3$
224 parameters	$(\Delta/\sigma)_{\rm max} = 0.002$
	$\Delta \rho_{\rm max} = 0.33 \ {\rm e} \ {\rm \AA}^{-3}$
	$\Delta \rho_{\rm min} = -0.22 \text{ e} \text{ Å}^{-3}$

Ta	ble	1
----	-----	---

Selected	geometric	parameters	(Å, °).
	0		× /	

C5-C6	1.342 (4)	C9-N2	1.381 (3)
C5-S1	1.747 (3)	C9-S2	1.748 (3)
C6-C7	1.442 (4)	C10-O1	1.228 (3)
C7-C8	1.376 (4)	C10-N2	1.414 (4)
C7-C10	1.431 (4)	C11-N2	1.455 (3)
C8-N1	1.364 (4)	C16-N3	1.381 (4)
C8-S1	1.727 (3)	C17-S2	1.801 (3)
C9-N1	1.298 (4)		
C8-S1-C5	90.46 (14)	C9-S2-C17	99.61 (16)



Figure 2

A partial packing diagram, showing a hydrogen-bonded (dashed lines) dimer. The suffix a corresponds to symmetry code i in Table 2.

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N3 - H3B \cdot \cdot \cdot N2$ N3 $H3A \cdot \cdot \cdot O1^{i}$	0.86(1)	2.51(4)	2.845 (4)	104 (3) 170 (2)
$N3-H3A\cdots OI^{*}$	0.86 (1)	2.29 (2)	3.139 (3)	170 (2)

Symmetry code: (i) -x + 1, -y, -z + 2.

H atoms attached to atom N3 were located in a difference Fourier map and refined with a restraint of N-H = 0.86 (1) Å. The displacement parameter $U_{iso}(H)$ was refined freely. Other H atoms were placed at calculated positions and treated as riding atoms, with C-H = 0.93-0.98 Å, and $U_{iso}(H) = 1.2U_{eq}(C)$, or $1.5U_{eq}(C)$ for CH₃.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 2001); software used to prepare material for publication: SHELXTL.

We gratefully acknowledge financial support of this work by the National Basic Research Programme of China (2003CB114400) and the National Natural Science Foundation of China (20372023 and 20102001).

References

- Bruker (2001). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Ding, M. W., Xu, S. Z. & Zhao, J. F. (2004). J. Org. Chem. 69, 8366-8371.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Sheldrick, G. M. (2001). SHELXTL. Version 5.0. Bruker AXS Inc., Madison, Wisconsin, USA.